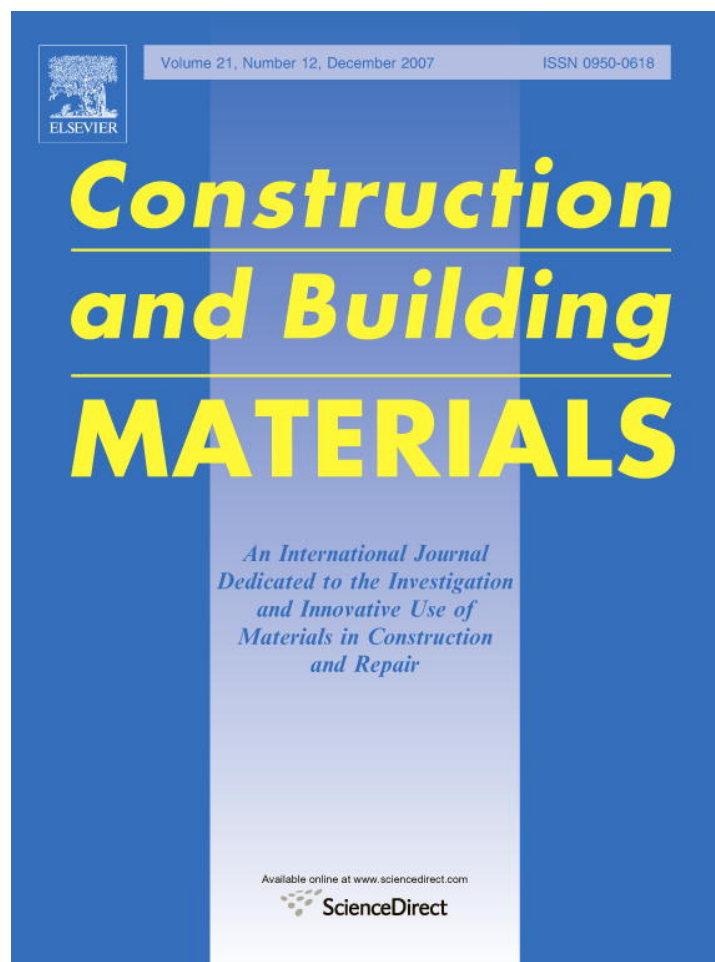


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Availability of tuffs from northeast of Turkey as natural pozzolan on cement, some chemical and mechanical relationships

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Abstract

At Northeast of Turkey, it is possible to find plenty of volcanic origin rocks and rich natural pozzolan beds. In this study, the pozzolanic activities of six types of different tuff samples taken from Trabzon and Bayburt regions (Northeast of Turkey) were examined according to the related standard and it was determined that the compressive strengths were varying between 6.7–11.0 MPa. In addition, the chemical compositions of these samples, except one, were consistent with the related standard. On the other hand, the results obtained from these studies were that increase in the proportion of SiO₂ in the pozzolan increases the pozzolanic activity.

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Keywords: Characterization methods; X-ray diffraction; Petrography; Compressive strength; Natural pozzolan

1. Introduction

Use of natural pozzolans, also known as “trass”, in cements or concrete systems make important effects on several properties of cement mortar such as strength, setting times, volume expansion and durability depending on its substitution ratio and its fineness [1–6]. In addition, since these materials enter the cement production after kiln process they also provide important economical and ecological benefits [7–9].

Natural pozzolans generally contain pyroclastic rocks. These rocks generally contain carbonate, clay and zeolite group minerals as filling/binding materials. Zeolitization is the transformation of the glassy structure in natural pozzolan to zeolite-group minerals with external effects. Zeolites can lose their water contents with the effect of heat. In other words, with the help of their channel shaped and

largely spaced ($d = 11.4 \text{ \AA}$) framed mineral structures they can easily take and release water ions [10]. These ions do not take part in the cage structure. One more important aspect is that zeolites can release alkali atoms and take in calcium and magnesium ions [11]. Therefore, when pyroclastics rich in zeolite minerals are mixed with slaked lime and water, they can form stable silicate minerals [12]. Besides zeolite, natural pozzolans can also include combinations of various silicate minerals such as quartz, feldspar, mica, hornblende, pyroxene, cristobalite, clay minerals, amorphous pumice, and glass shards. In general, a good pozzolan has low quantities of clay minerals, low quantities of alkali feldspar, high quantities of zeolite minerals, and volcanic glass. In addition, it should exhibit high porosity and specific surface area [13]. Turkey is rich in natural pozzolans. Almost 155,000 km² of the country is covered by Tertiary- and Quaternary-age volcanic rocks, among which tuffs occupy important volumes. Although there are many geological investigations on these volcanics and their potential as natural pozzolans is not well established [14]. Former studies show that there are zeolites as the most common secondary mineral in the volcanic rocks of the

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Trabzon region, especially those containing feldspars. Besides these, widely found clay has completed its weathering period [15].

Pozzolanic activity can be defined as the reactivity capacity with $\text{Ca}(\text{OH})_2$ and hardening capacity in humid medium [16]. Although some other properties have effects on the pozzolanic activity, the references states that the pozzolan's strength is especially affected by the chemical composition, fineness and the amount of its glassy phase [17]. In the German standards (DIN 51043) the chemical composition of natural pozzolans is stated as 50–67% SiO_2 , 14–20% Al_2O_3 , 2–5% Fe_2O_3 , less than 10% $\text{CaO}+\text{MgO}$ and 3–8% $\text{Na}_2\text{O}+\text{K}_2\text{O}$ [18]. The restrictions in the Turkish standard (TS 25 [19]) are given as a sum of the main ingredients similar to ASTM C311 [20].

The objective of this study is determining the availability of the tuffs taken from unused new fields of the North-east of Turkey (Trabzon and Bayburt district) as natural pozzolan on cement and making the connections between pozzolanic activities of the materials and their petrographic-chemical compositions. In addition, the correctness of restrictions in the standard, TS 25 [19], will be discussed in a way.

2. Materials and experimental methods

Locations of samples were carried out, by examining geological maps of Trabzon and Bayburt (cities in North-east of Turkey), and 42 samples were chosen for studies. Mineralogical and petrographical properties of the natural pozzolan samples were identified under the polarizing microscope by using their thin sections (Fig. 1) and X-ray powder diffraction (XRD) analysis of the powdered bulk

samples were carried out (Fig. 2). Major element analysis of the samples comprising SiO_2 , Al_2O_3 , Fe_2O_3 , MgO , CaO , Na_2O and K_2O were carried out by means of wet chemical analysis. The pozzolanic activities, the behavior of pozzolan in mortar, of these six samples were determined according to TS 25 [19]. According to this standard, at least three mortar specimens were prepared by mixing slaked lime, standard sand, and natural pozzolan (tuff) having the weights shown in Table 1. Immediately after molding, the mortar specimens were covered to prevent evaporation and cured in a moist environment (at $21\pm 1^\circ\text{C}$) for 24 h. The specimens were tested for their 7th day flexural and compressive strength according to EN 196–1 by the “Rilem–Cembureau Method” [21].

3. Results

3.1. Mineralogical and petrographical evaluations

The results of the carried out thin section and XRD investigations, as summarized in Table 2, the samples, except one (P; basanite), were characterized as dacitic and andesitic tuffs. These samples generally contain minerals having high ratio of SiO_2 such as quartz, plagioclase, amphibole, chlorite, illite, sanidine, clinopliolite and pyroxene. Besides, there are also zeolitization in secondary inter-gaps of the samples.

SiO_2 component can exist in both crystalline and non-crystalline form minerals. That the non-crystalline minerals like zeolites have high pozzolanic activity is well known. However, besides this, crystalline and mix-crystalline minerals also contribute pozzolanic activity. The high pozzolanic activity results in this study also prove this.

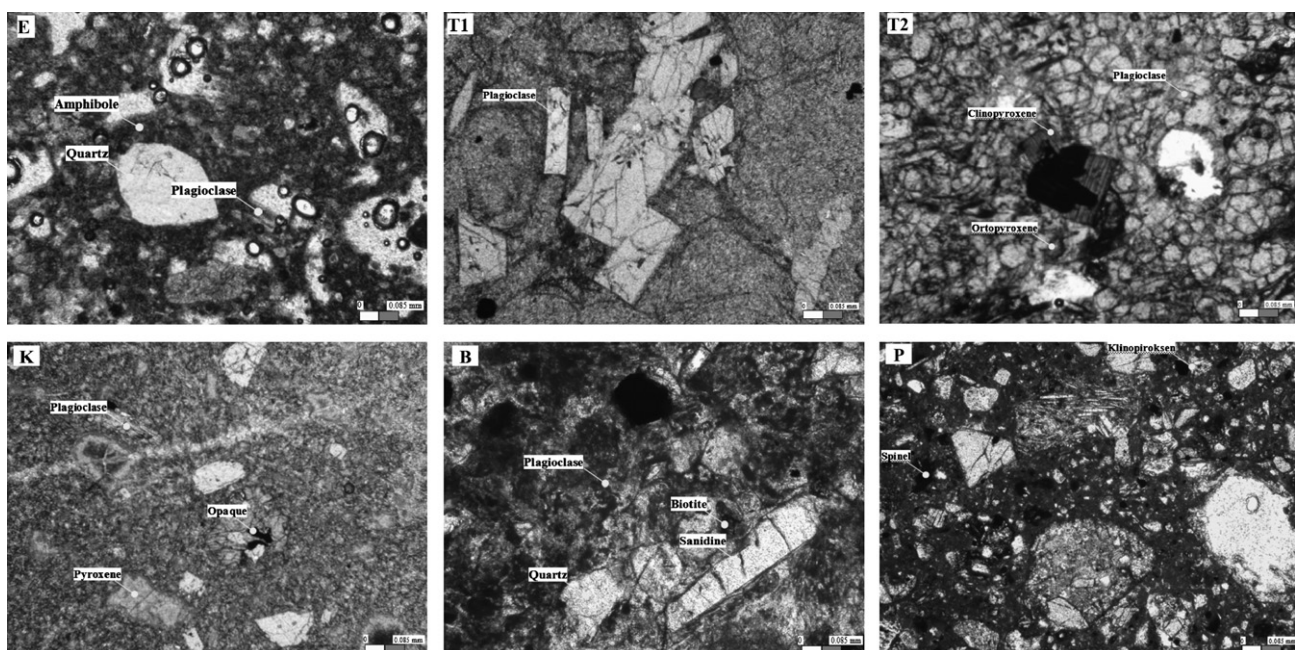


Fig. 1. Thin sections of the pozzolan.

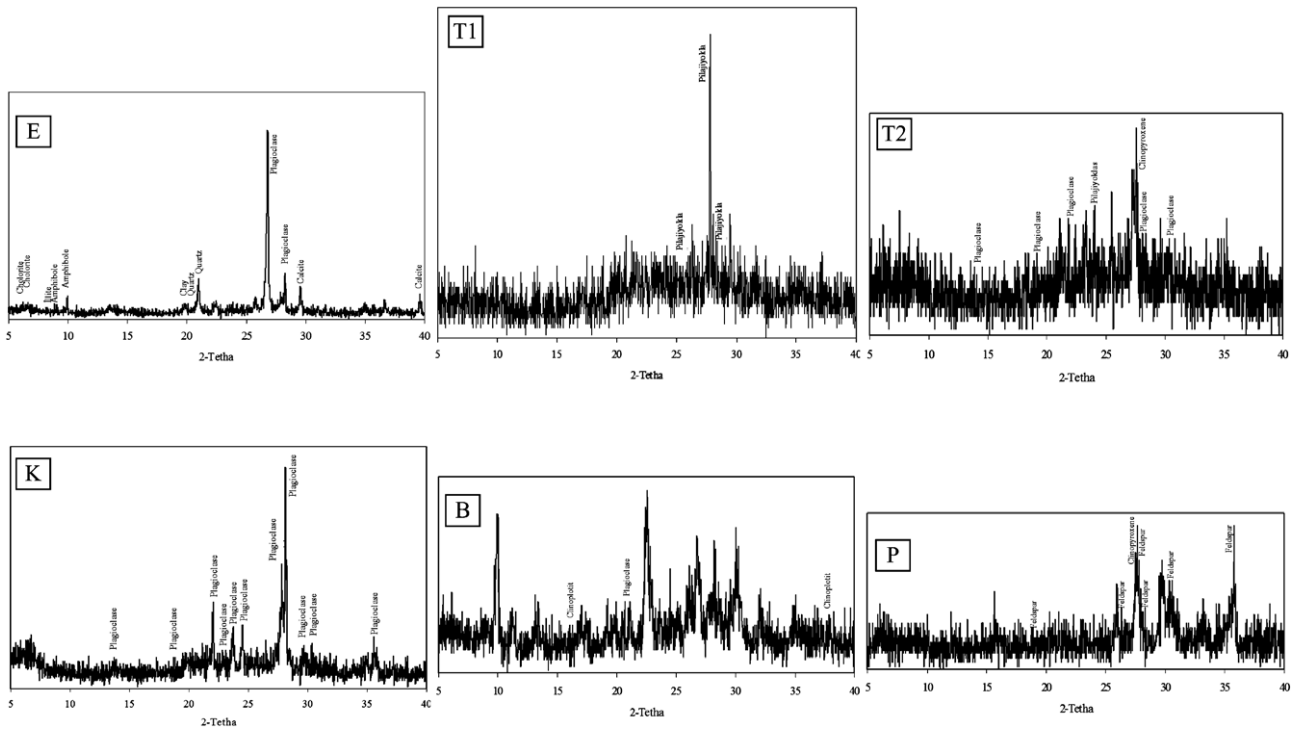


Fig. 2. XRD of pozzolans.

Table 1
Proportions of the ingredients of mortar specimens according to TS 25 standard

| Material | Weight (in gram) |
|---------------------------------|------------------|
| Slaked lime Ca(OH) ₂ | 150 |
| Pozzolan | 2×150×N = T |
| Standard sand | 1350 |
| Water | 0.5×(150 + T) |

N = factor obtained by dividing density of tuff by the density of lime.

Table 2
Petrographic properties of the pozzolans

| Sample | Characteristic | Mineral content |
|--------|----------------|---|
| E | Dacitic tuff | Quartz, plagioclase, amphibole, chlorite, illite, calcite |
| T1 | Andesitic tuff | Plagioclase |
| T2 | Andesitic tuff | Plagioclase, Orthopyroxene, clinopyroxene |
| K | Andesitic tuff | Plagioclase, pyroxene, opaque |
| B | Dacitic tuff | Quartz, plagioclase, sanidine, biotite, clinoptilolite |
| P | Basanite | Clinopyroxene, spinel, leucite |

3.2. Chemistry of samples

According to the results of the performed chemical analyses, it was determined that the samples have high ratio of SiO₂. The sources of this component were given above. According to related standard [19], some chemical properties are required from the materials that are to be used as pozzolan. The results of the chemical analysis of the pozzo-

lan samples (Table 3) and their consistency with this standard (Table 4) are shown briefly. Besides, the physical properties of the samples were given in Table 5.

As seen in Table 4, the samples with the exception of the P sample are within the limits of TS 25 standard. Because the sum of SiO₂ + Al₂O₃ + Fe₂O₃ is 66.43% <70% (by mass) for sample P. This sample is not appropriate for using as pozzolan in cement in the view of chemical properties. Besides, it was seen that MgO ratio was only a little below the limit. Although sample P did not fulfill the required properties, it was also subjected to the pozzolanic activity tests and according to these results, it was appropriate to standard exceedingly (Table 6).

3.3. Pozzolanic activities and their relationships with chemical compositions

Pozzolanic activity experiment results (Table 6) of prepared pozzolan samples are above the minimum strength values determined by the standards. According to the results, all six samples are mechanically suitable for usage in the cement as pozzolan. Additionally, although sample P does not fulfill the chemical requirements due to the scarcity of Si, Al, Fe oxide rates, it is suitable as high as the mechanical properties expressed in the TS 25 standard [19].

When the relationship between SiO₂, Al₂O₃, Fe₂O₃, CaO and the other components, which are the main ingredients of pozzolan, and pozzolanic activity is concerned, the probability of having an idea about the strength of the pozzolan by looking at its chemical composition can be possible.

Table 3
Chemical properties of the pozzolans

| Sample | Insoluble residue | Chemical analysis (% _{mass}) | | | | | | | | | |
|--------|-------------------|--|--------------------------------|--------------------------------|-------|------|-----------------|-------|-------------------|------------------|-------|
| | | SiO ₂ | Al ₂ O ₃ | Fe ₂ O ₃ | CaO | MgO | SO ₃ | LOI | Na ₂ O | K ₂ O | Total |
| E | 77.24 | 70.89 | 9.08 | 2.96 | 5.40 | 0.62 | – | 7.23 | 1.11 | 1.92 | 99.21 |
| T1 | 83.65 | 67.96 | 13.73 | 2.57 | 3.24 | 1.12 | 0.51 | 6.37 | 2.54 | 1.40 | 99.44 |
| T2 | 85.75 | 67.93 | 4.43 | 1.26 | 2.28 | 0.61 | 0.48 | 4.61 | 4.48 | 2.92 | 88.39 |
| K | 79.90 | 61.01 | 15.10 | 5.35 | 5.16 | 1.74 | 0.71 | 3.88 | 3.76 | 2.88 | 99.59 |
| B | 55.33 | 60.14 | 13.17 | 4.35 | 4.83 | 2.34 | 0.17 | 10.71 | 0.71 | 3.08 | 99.50 |
| P | 54.63 | 43.10 | 13.77 | 9.56 | 13.52 | 4.80 | – | 6.00 | 2.29 | 2.02 | 95.06 |

Table 4
The comparison of the pozzolans with TS 25 in point of chemical composition

| | TS 25 | E | T1 | T2 | K | B | P |
|--|----------|-------|-------|-------|-------|-------|-------|
| SiO ₂ + Al ₂ O ₃ + Fe ₂ O ₃ (%) | Min. %70 | 82.93 | 84.26 | 73.62 | 81.46 | 77.76 | 66.43 |
| MgO (%) | Max. %5 | 0.62 | 1.12 | 0.61 | 1.74 | 2.34 | 4.80 |
| SO ₃ (%) | Max. %3 | – | 0.51 | 0.48 | 0.71 | 0.17 | – |

Table 5
The physical properties of pozzolan samples prepared for experiment

| Sample | Remain 200 µm sieve (%) | Remain 90 µm sieve (%) | Specific surface cm ² /g | Specific mass g/cm ³ | Unit mass g/cm ³ | Grinding time (minute) |
|--------|-------------------------|------------------------|-------------------------------------|---------------------------------|-----------------------------|------------------------|
| B | 0.1 | 4.6 | 6226 | 2.25 | 658 | 35 |
| K | 0.1 | 3.3 | 5286 | 2.40 | 747 | 30 |
| T1 | 0.1 | 3.0 | 4153 | 2.29 | 739 | 30 |
| T2 | 0.1 | 3.0 | 4869 | 2.40 | 765 | 25 |
| E | 0.6 | 4.4 | 5908 | 2.49 | – | 45 |
| P | 0.3 | 2.7 | 5887 | 2.71 | – | 40 |

Table 6
The mechanical properties of pozzolan samples and comparison with TS 25 standard

| Sample | 7th day flexural strength (MPa) | Min. according to TS 25 (MPa) | 7th day compressive strength (MPa) | Min. according to TS 25 (MPa) |
|--------|---------------------------------|-------------------------------|------------------------------------|-------------------------------|
| B | 3.02 | 1.00 | 8.30 | 4.00 |
| K | 1.90 | | 6.70 | |
| T1 | 3.96 | | 10.10 | |
| T2 | 3.14 | | 9.10 | |
| E | 4.45 | | 11.00 | |
| P | 2.86 | | 9.00 | |

According to TS 25 standard, SiO₂, Al₂O₃, Fe₂O₃ sum should be at least 70% (by mass) [19]. This shows that, in addition to durability contribution, these components are the main effects at actuating the pozzolanic activity. The validness of the present claim is discussed with the help of graphics and relations by taking into consideration the effects of each element separately and totally. In addition, these investigations are performed for also MgO and K₂O.

In the graphics (Fig. 3), the discontinuous lines represent the linear relation of the six samples, whereas the continuous lines represent the relations obtained by not considering the one sample—indicated with hollow symbol—, which does not have close values with the others. Namely, the relations in the graphics are deduced from the closest five samples.

As a result of the comparisons made, highly correlated (88%) relation between the SiO₂ rate and compressive strength was established, and as seen from the graph

(Fig. 3) it was determined that pozzolanic activity increases with the increase in SiO₂ rate. It was also thought that not only in non-crystalline form SiO₂ but also in crystalline form SiO₂ was increased the pozzolanic activity. The increase in the Al₂O₃ decreases the compressive strength. Also it is seen from the graph that the correlation coefficient is quite high (79%). Similarly, compressive strength decreased along with an increase in the Fe₂O₃ composition. Therefore, an increase in the sum of Al₂O₃ and Fe₂O₃ rates will decrease the compressive strength. As just seen, although an increase in the sum of these three elements increase the strength, this increase is in fact caused by SiO₂. This should be accepted as a finding that assist to reinvestigate the stipulation of the Al₂O₃% + SiO₂% + Fe₂O₃% >70% limit mentioned in TS 25. Therefore, the meaningfulness of the West European standard mentioned above [18], which was made a separate limitation about SiO₂ and other component rate, is seen. On the other hand,

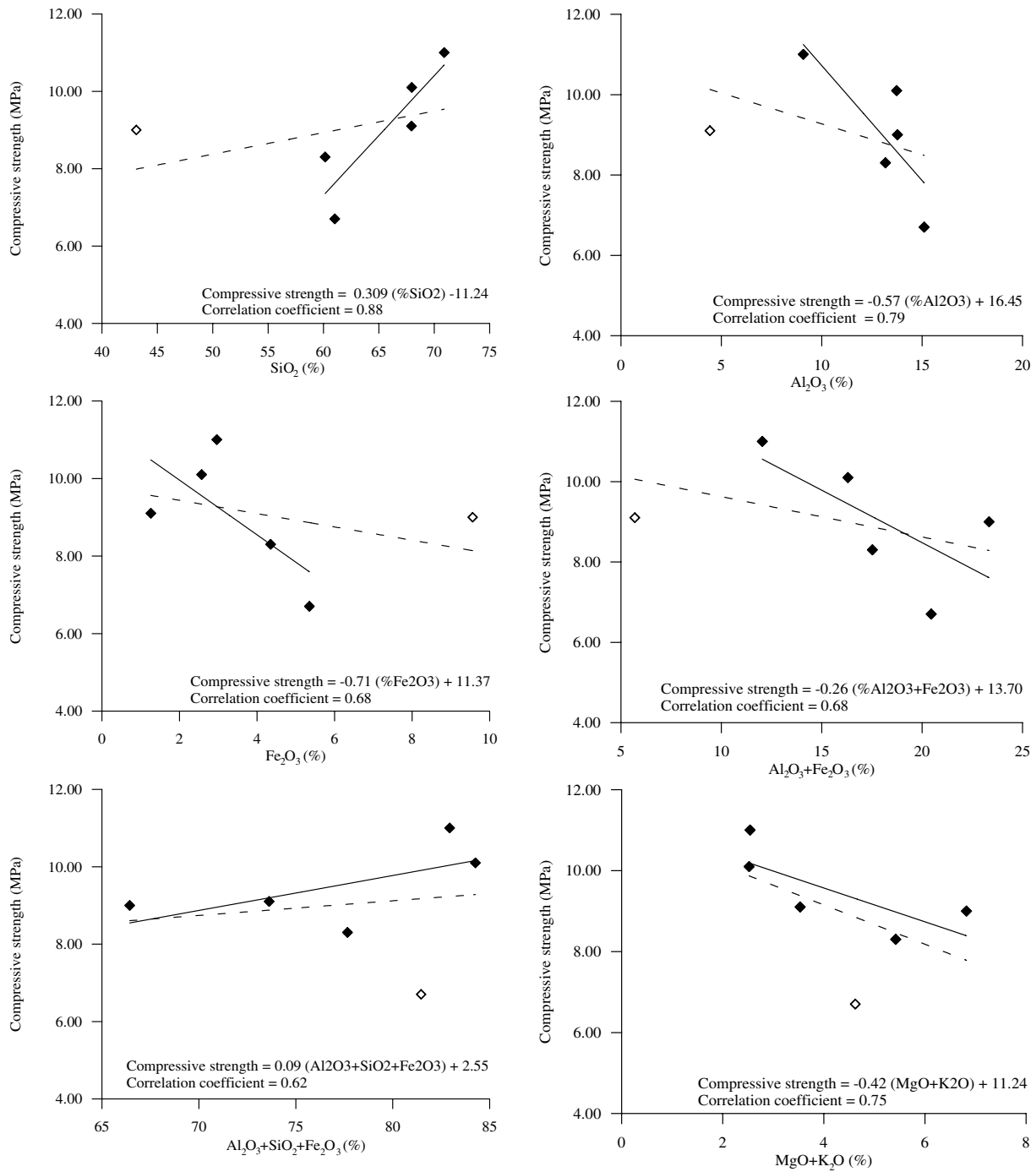


Fig. 3. The relations between components of pozzolan and compressive strength.

when the MgO and K_2O rate increased as a total, a decrease in the compressive strength is seen (Fig. 3). Here again, the relation is appearing with a high correlation coefficient (75%).

The results show that the most important component increasing the pozzolanic activity of a pozzolan is SiO_2 oxide. Al_2O_3 , Fe_2O_3 , MgO and K_2O oxides decrease the pozzolanic activity. It is possible to say that there are two reasons of increasing strengths of pozzolans, and so

cements, by increasing SiO_2 ratio: firstly, the minerals that contain SiO_2 can be ground finely and so it can fill micro pores. The second is that the capable of binding $\text{Ca}(\text{OH})_2$ of SiO_2 is higher than the others' (Al_2O_3 , Fe_2O_3). According to latter, this reaction contributes to form of calcium-silicate-hydrate in shorter time. There are also standards as EN 196–5 about the determination of these properties [22]. Therefore, it is possible to say that the rocks those are used as pozzolan should be high proportion of SiO_2 .

4. Conclusions

As seen from this study, it possible to easily find rich natural pozzolan resources in Northeast of Turkey (Trabzon–Bayburt district). The rocks that were taken from this region and investigated were volcanic origin and they showed pozzolanic property. They consisted of tuffs, except one, and they mostly fulfilled the TS 25 standard requirements for both pozzolanic activity and chemical composition. With this result, by using these natural pozzolan resources in cement production, important amounts of heat saving can be possible and this can help stopping global warming and excess CO₂ emissions, which are the common ecological problems of humanity.

Another result obtained from this study, the pozzolanic activities of natural pozzolans and hence their compressive strengths are directly proportional with the increase in SiO₂ ratio and inversely proportional to Al₂O₃ and Fe₂O₃ ratios. Again it is seen that increase in the MgO and K₂O ratios decrease the pozzolanic activity. According to this, it is once again seen that the most important component of pozzolan is SiO₂ and it can provide contributions to pozzolanic activity in both crystalline form and non-crystalline form.

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